

## Phthalonitrile Conductive Polymer Chemical Vapor Sensors

John F. Giuliani\* and Teddy M. Keller\*\*

(\*Code 6170, \*\*Code 6127)  
Chemistry Division  
Naval Research Laboratory  
Washington, DC 20375-5000  
U.S.A.

(Received April 17, 1989; accepted for publication July 25, 1989)

**Key words:** phthalonitrile conductive polymer, chemical vapor sensor for ammonia and acetone.

An intrinsically stable, highly conductive polymeric chemical vapor sensor has been fabricated from an aromatic, diether-linked phthalonitrile resin. Preliminary chemical vapor studies show that this pyrolyzed semiconductive material responds reversibly and somewhat selectively to ammonia and acetone from among a number of vapors tested. Electrical resistance at room temperature is increased by a few percent within the first 10 to 30 seconds of exposure to these vapors in a dynamic equilibrium nitrogen carrier/vapor flow dilution system. A concentration-dependent electrical response study was carried out in which the polymer was exposed to a range of dilution concentrations of dry ammonia gas at room temperature. Sample voltage changes between 23 mV and 14 mV were measured over ammonia/nitrogen dilution concentrations from 10,000 ppm to approximately 45 ppm.

### 1. Introduction

Electrically conductive polymeric materials are finding wide use in the area of integrated solid-state electronics in which these materials are gradually replacing metals and inorganic semiconductors. The scope of applications include batteries, EMI shielding, biomedical membranes, ion-exchange devices, ionization detectors, chemical vapor sensors and indicators.<sup>(1)</sup>